

Name: _____

Form A

Formula

- 1 $W(CN)_2$
- 2 $Cu(HCO_3)_2$
- 3 $W(CO_3)_3$
- 4 $Cd(HSO_3)_2$
- 5 PBr_3
- 6 $Pb_3(PO_3)_2$
- 7 NiF_3
- 8 Pb_3N_4
- 9 $W_3(PO_4)_4$
- 10 H_2SO_4
- 11 $Cr(ClO_2)_2$
- 12 $HgCO_3$
- 13 UPO_4
- 14 Na_2CO_3
- 15 K_2SO_4
- 16 XeF_6
- 17 SiO_2
- 18 $Mn_2(SO_4)_7$
- 19 H_3PO_4
- 20 $NiCl_2$
- 21 $Co(ClO_2)_3$
- 22 XeF_4

- 23 $Be(ClO_4)_2$
- 24 $Mn(ClO_4)_7$
- 25 CH_3COOH
- 26 CCl_3
- 27 $Ti(NO_3)_2$
- 28 $CaSO_4$
- 29 $Pb(H_2PO_4)_2$
- 30 $Mn(NO_3)_6$
- 31 BrF
- 32 $Cu(ClO)_2$
- 33 $FeCl_6$
- 34 P_2I_4
- 35 P_4S_3
- 36 HCN
- 37 HF
- 38 $RaCO_3$
- 39 RbI
- 40 $CrCl_6$

Name

- 41 silicon tetrabromide
- 42 cesium chlorite
- 43 perchloric acid
- 44 chloric acid
- 45 iodine monochloride
- 46 disulfur dichloride
- 47 cobalt(III) sulfate

48	chromium(II) fluoride	76	chromium(II) sulfite
49	carbonic acid	77	nickel(III) hydride
50	magnesium cyanide	78	gold(III) hypochlorite
51	tin(II) hypochlorite	79	hydrobromic acid
52	xenon tetroxide	80	hydroiodic acid
53	sulfur hexafluoride		
54	hydrosulfuric acid		
55	titanium(IV) hypochlorite		
56	manganese(II) phosphate		
57	hypochlorous acid		
58	barium bicarbonate		
59	chlorous acid		
60	nitrogen tribromide		
61	phosphorous acid		
62	dinitrogen tetroxide		
63	nitric acid		
64	carbon tetrachloride		
65	thiosulfuric acid		
66	hydrochloric acid		
67	strontium sulfite		
68	nitrous acid		
69	gold(III) nitrite		
70	radium fluoride		
71	sulfur tetrafluoride		
72	uranium(IV) chlorite		
73	uranium(IV) hydrogen phosphate		
74	nickel(III) sulfide		
75	xenon difluoride		